



eCRAM computer algorithm for implementation of the charge ratio analysis method to deconvolute electrospray ionization mass spectra

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ABSTRACT

A computer program (eCRAM) has been developed for automated processing of electrospray mass spectra based on the charge ratio analysis method. The eCRAM algorithm deconvolutes electrospray mass spectra solely from the ratio of mass-to-charge (m/z) values of multiply charged ions. The program first determines the ion charge by correlating the ratio of m/z values for any two (i.e., consecutive or non-consecutive) multiply charged ions to the unique ratios of two integers. The mass, and subsequently the identity of the charge carrying species, is further determined from m/z values and charge states of any two ions. For the interpretation of high-resolution electrospray mass spectra, eCRAM correlates isotopic peaks that share the same isotopic compositions. This process is also performed through charge ratio analysis after correcting the multiply charged ions to their lowest common ion charge. The application of eCRAM algorithm has been demonstrated with theoretical mass-to-charge ratios for proteins lysozyme and carbonic anhydrase, as well as experimental data for both low and high-resolution FT-ICR electrospray mass spectra of a range of proteins (ubiquitin, cytochrome c, transthyretin, lysozyme and calmodulin). This also included the simulated data for mixtures by combining experimental data for ubiquitin, cytochrome c and transthyretin.

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1. Introduction

Biopolymers are routinely analyzed by electrospray ionization (ESI) with high-mass accuracy when combined with Fourier transform ion cyclotron resonance mass spectrometry (FT-ICR MS) [1,2]. Computer algorithms have been developed to expedite the analysis of electrospray mass spectra and to accurately translate mass-to-charge (m/z) ratios of highly charged ions to zero-charge molecular mass values. For low-resolution mass spectra, where isotope peaks of multiply charged ions are not resolved, algorithms were developed by assuming the nature of the charge carrying species or considering only a limited set of charge carrying species [3,4]. These algorithms had other limitations of producing artefact peaks that were greatly reduced by incorporating maximum-entropy based or multiplicative correlation algorithms [5,6].

With the high-resolving power of ion cyclotron resonance (ICR) mass analyzers, multiply charged ions generated by ESI could be resolved to their isotopic compositions affording mass accuracies in the part-per-million (ppm) range, provided both the ion charge (z) and the assignment of an ion's isotopic composition was accurately determined. In the case of high-resolution electrospray mass

spectra, the ion charge can be derived directly from the reciprocal of the mass-to-charge separation between adjacent isotopic peaks ($1/\Delta m/z$) for any multiply charged ion [7,8]. Although the isotope spacing method is direct, the complexity of overlapping isotope peaks for mixtures and the addition of spectral noise may result in inaccurate ion charge determination. Furthermore, for high charge states ions, distinguishing $1/z$ and $1/(z+1)$ would require mass accuracies of a few ppm that is not routinely possible. To overcome some of these limitations, pattern recognition techniques were combined with the isotope spacing method in Zscore [9] and THRASH [10] algorithms for automated charge state determination of FT-ICR mass spectra. For example, the THRASH algorithm incorporated matching the experimental abundances with theoretical isotopic distributions based on the model amino acid *averagine* ($C_{4.938}H_{7.7583}N_{1.3577}O_{1.4773}S_{0.0417}$) [8], which restricts its application to a specific group of compounds and elemental compositions.

Other algorithms developed in recent years, AID-MS [11] and PTFT [12], work on the basis of subtractive peak finding routines to locate possible isotopic clusters in the spectrum. The former algorithm shares similarities with the THRASH and relies on the isotope spacing method as well as correlating the experimental isotopic distributions with theoretical patterns (i.e., known elemental compositions). The AID-MS algorithm [11] requires the use of *averagine* for modeling isotopic distributions, while the PTFT subtracts the peak intensities in the frequency domain [12].

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We have developed the charge ratio analysis method (CRAM) for ESI analysis of biopolymers [13,14]. The unique feature of the CRAM is that charge states of ions are identified *purely* from the ratios of m/z values of different multiply charged ions and by correlating these values to unique ratios of two integers that subsequently identify charge of ions. This approach is therefore *independent* of the charge carrying species and as such the nature of charge carrying species is not kept *uniform* for all ions in a mass spectrum. For high-resolution data, the CRAM process could also correlate the isotopic peaks of different multiply charged ions that share the same isotopic compositions [14]. This isotopic correlation step can subsequently result in a more accurate mass determination, particularly for sample mixtures with overlapping cluster peaks and for low signal-to-noise data. In addition, the application of isotopic peak correlation within the CRAM can be valuable for mass spectrometric quantification approaches that rely on correlation of accurate isotopic peaks [15–19]. Note that the CRAM does not require a prior knowledge of the elemental composition of a molecule, and as such does not rely at all on correlating experimental isotopic patterns with the theoretical patterns (i.e., known compositions), and therefore CRAM could be applied to mass spectral data for a range of compounds (i.e., including unspecified compositions). The application of the CRAM algorithm (eCRAM) for analysis of both the low and high-resolution ESI data is described, and the algorithm is demonstrated by processing data for several proteins.

2. Experimental

Electrospray mass spectra were recorded on a 4.7 Tesla magnet (APEX, Bruker Daltonics, Billerica, MA, USA) mass spectrometer in the positive ion mode and spectra were processed with 512k or 1M data points. Theoretical isotopic distributions were produced from the isotopic distribution utility of the Xmass software (Bruker Daltonics, Billerica, MA, USA). Protein samples were obtained from Sigma Chemicals (St. Louis, MO, USA), and were used without further purification, and solutions were prepared at a concentration range of 1–5 μM in 50:50 water and methanol or acetonitrile containing 2–4% acetic acid or 0.1% TFA. The sample of transthyretin (TTR) was provided in a buffer of 400 mM sodium phosphate from the Scripps Research Institute (La Jolla, CA, USA) [20], and a C₁₈ Sep-Pak (Waters Corporation, MA, USA) was used for desalting. Solutions were infused at a rate of between 3 and 5 $\mu\text{L}/\text{min}$ with an electrospray needle voltage in the range of 4.2–4.5 kV. Spectra were mass calibrated with the most abundant isotopic peak of angiotensin-1 (2+ and 3+ ions) and ubiquitin (7+ to 13+) as external or internal calibration ions.

3. Results and discussions

3.1. Theoretical basis of the CRAM

The theoretical basis of the CRAM was fully described earlier [11,12], and is briefly explained here. The mass-to-charge (m/z) values for two multiply charged ions, i and j , originating from the same compound with a molecular mass (M) can be represented as $(R_z)_i$ and $(R_z)_j$, respectively. These multiply charged ions of charge (z) would correspond to the addition or abstraction of a charge carrying species (m_A), where $R_z = (M \pm zm_A)/z$. The ratio of m/z values for two ions, i and j , can then be represented by Eq. (1):

$$\frac{(R_z)_i}{(R_z)_j} = \frac{z_j(M \pm z_i m_A)}{z_i(M \pm z_j m_A)} \quad (1)$$

The CRAM approach makes an assumption that $M > z_i m_A$ or $z_j m_A$, and therefore Eq. (1) is simplified to Eq. (2):

$$\frac{(R_z)_i}{(R_z)_j} = \frac{z_j}{z_i} \quad (2)$$

Thus from the ratio of m/z values of any two multiply charged ions, the inverse ratio of their two ion charges can be calculated. The unique property of the ratio of two integers is the basis of the CRAM where by correlating the ratio of m/z values of two multiply charged ions to the unique ratio of two integers, the charge states of ions are identified without *a priori* knowledge or assumption of the nature of the charge carrying species. The purely mathematical basis of the CRAM makes it applicable to mass spectral analyses of proteins, oligonucleotides or carbohydrates with a wide range of compositions (i.e., proteins and non-proteins).

The mass (m_A) and subsequent identity of the charge carrying species is determined by the CRAM according to Eq. (3):

$$((R_z)_i z_i - (R_z)_j z_j) = \pm m_A (z_i - z_j) \quad (3)$$

For high-resolution ESI data, in order to correlate two isotopic peaks across two different multiply charged ion distributions of charge z_i and z_j where $z_i > z_j$ and $i = j + n$, the m/z value for $(R_z)_i$ in Eqs. (1) to (3) need to be corrected for the additional mass of the charge carrying species by substituting it with $(R_z)_i - n(m_A/z_i)$ [14]. This correction factor is important for high-resolution accurate mass determination, where Eq. (4) can be used to correlate isotopic peaks that share a common isotopic composition:

$$\frac{(R_z)_i - n(m_A/z_i)}{(R_z)_j} = \frac{z_j}{z_i} \quad (4)$$

The value for n represents the difference in charge of the multiply charged ions that are being correlated ($n = i - j$). For ions of charge 13 and 10, the value for n is 3.

The CRAM processes electrospray mass spectra without a theoretical upper mass limit, and was previously demonstrated to calculate charge states more accurately, on the order of 100–1000 fold, in comparison to the isotope spacing method [14]. Consider for example, human carbonic anhydrase II (Protein Data Bank entry 1ca2) with the elemental composition of C₁₃₂₄H₂₀₁₉N₃₅₆O₃₈₃S₂, a monoisotopic mass of 29097.88961 Da and m/z values 2910.79679, 2646.27052 and 2425.83196 corresponding to additions of +10 to +12 protons (i.e., charges). The ratio of 2910.79679/2646.27052 is 1.09996 that is within 4×10^{-5} of 1.1 derived from dividing 11 by 10, or the ratio of 2910.79679/2425.83196 is 1.19992, which is within 8×10^{-5} of 1.2 derived from dividing 12 by 10. After applying the correction factor (Eq. (4)), and adjusting the m/z values to a common charge state of 10 (i.e., 2910.79679, 2646.27052 and 2425.83196), the ratio of 2910.79679/2646.27052 is 1.1 that is the same as dividing 11 by 10.

3.2. Computational basis of eCRAM algorithm

The CRAM algorithm (eCRAM) has been implemented using the Perl programming language for flexibility, in combination with Unix command line utilities for efficient sorting and stream editing. The program accepts, as input a comma separated file containing peak positions (m/z) and relative peak intensities. Spectral peaks are sorted automatically into increasing m/z values.

The program approaches the analysis by first populating the space of all possible solutions and then efficiently identifying physically realistic solutions within that space. The solution space is four-dimensional since each pair of spectral peaks (i and j) is combined with candidate charge values (k and l). The term “row”

will be used to refer to each unique combination of (*i*, *j*, *k*, and *l*) in the solution space. The physically realistic solutions can be drawn from the best rows (i.e., those with the smallest errors between their ratio of *m/z* values and their ratio of charge values).

For each row, that is a pair of spectral peaks, *i* and *j*, and candidate charges, *k*, and *l*, the program computes the experimental ratio of *m/z* values for the two peaks, as well as with the theoretical charge ratio of the candidate charge values. The amount of discrepancy between experimental and theoretical charge ratios can be

A: Theoretical Mass List for Lysozyme

Row Number	M/Z	Relative Intensity
0	1101.403480	100.0000
1	1123.385380	100.0000
2	1139.494080	100.0000
3	1193.103120	100.0000
4	1215.085020	100.0000
5	1231.193720	100.0000
6	1301.475420	100.0000
7	1323.457320	100.0000
8	1339.566020	100.0000
9	1431.522180	100.0000
10	1453.504080	100.0000
11	1469.612780	100.0000

```
Read 12 rows of input data from file ..
Normalise input data
Evaluated maximum intensity as 100.0000 for the 12 data lines of input
Looks like LOW resolution data with 1 bunch of spectral peaks
```

B: Best 50 Rows on Difference Value (Δ_{ijkl})

Diff	Peaks	Charges	RZa	RZb	delta_m	Ma	Mb	dM / dZ	{RowNum}		
1	0	3	13	12	14318.2452	14317.2374	1.0078	14318.2452	14317.2374	1.0078	{ 0}
1	3	6	12	11	14317.2374	14316.2296	1.0078	14317.2374	14316.2296	1.0078	{ 1}
1	6	9	11	10	14316.2296	14315.2218	1.0078	14316.2296	14315.2218	1.0078	{ 2}
1	0	6	13	11	14318.2452	14316.2296	1.0078	14318.2452	14316.2296	1.0078	{ 3}
1	3	9	12	10	14317.2374	14315.2218	1.0078	14317.2374	14315.2218	1.0078	{ 4}
1	3	9	18	15	21475.8562	21472.8327	1.0078	21475.8562	21472.8327	1.0078	{ 5}
2	0	5	19	17	20926.6661	20930.2932	1.8136	20926.6661	20930.2932	1.8136	{ 6}
2	0	9	13	10	14318.2452	14315.2218	1.0078	14318.2452	14315.2218	1.0078	{ 7}
3	4	6	15	14	18226.2753	18220.6559	5.6194	18226.2753	18220.6559	5.6194	{ 8}
3	2	4	16	15	18231.9053	18226.2753	5.6300	18231.9053	18226.2753	5.6300	{ 9}
4	1	11	17	13	19097.5515	19104.9661	1.8537	19097.5515	19104.9661	1.8537	{ 10}
4	1	3	17	16	19097.5515	19089.6499	7.9015	19097.5515	19089.6499	7.9015	{ 11}
6	7	11	10	9	13234.5732	13226.5150	8.0582	13234.5732	13226.5150	8.0582	{ 12}
6	2	6	8	7	9115.9526	9110.3279	5.6247	9115.9526	9110.3279	5.6247	{ 13}
6	2	6	16	14	18231.9053	18220.6559	5.6247	18231.9053	18220.6559	5.6247	{ 14}
7	0	11	12	9	13216.8418	13226.5150	3.2244	13216.8418	13226.5150	3.2244	{ 15}
7	0	11	16	12	17622.4557	17635.3534	3.2244	17622.4557	17635.3534	3.2244	{ 16}
7	6	10	19	17	24728.0330	24709.5694	9.2318	24728.0330	24709.5694	9.2318	{ 17}
8	3	11	16	13	19089.6499	19104.9661	5.1054	19089.6499	19104.9661	5.1054	{ 18}
11	5	10	13	11	16005.5184	15988.5449	8.4867	16005.5184	15988.5449	8.4867	{ 19}
12	1	9	14	11	15727.3953	15746.7440	6.4496	15727.3953	15746.7440	6.4496	{ 20}
13	0	7	12	10	13216.8418	13234.5732	8.8657	13216.8418	13234.5732	8.8657	{ 21}
13	0	7	18	15	19825.2626	19851.8598	8.8657	19825.2626	19851.8598	8.8657	{ 22}
14	5	6	19	18	23392.6807	23426.5576	33.8769	23392.6807	23426.5576	33.8769	{ 23}
16	7	9	13	12	17204.9452	17178.2662	26.6790	17204.9452	17178.2662	26.6790	{ 24}
16	1	4	13	12	14604.0099	14581.0202	22.9897	14604.0099	14581.0202	22.9897	{ 25}
16	4	7	12	11	14581.0202	14558.0305	22.9897	14581.0202	14558.0305	22.9897	{ 26}
16	7	10	11	10	14558.0305	14535.0408	22.9897	14558.0305	14535.0408	22.9897	{ 27}
16	8	10	13	12	17414.3583	17442.0490	27.6907	17414.3583	17442.0490	27.6907	{ 28}
16	0	8	17	14	18723.8592	18753.9243	10.0217	18723.8592	18753.9243	10.0217	{ 29}
16	5	6	18	17	22161.4870	22125.0821	36.4048	22161.4870	22125.0821	36.4048	{ 30}
17	3	7	10	9	11931.0312	11911.1159	19.9153	11931.0312	11911.1159	19.9153	{ 31}
18	5	7	14	13	17236.7121	17204.9452	31.7669	17236.7121	17204.9452	31.7669	{ 32}
19	8	9	16	15	21433.0563	21472.8327	39.7764	21433.0563	21472.8327	39.7764	{ 33}
20	3	8	9	8	10737.9281	10716.5282	21.3999	10737.9281	10716.5282	21.3999	{ 34}
20	3	8	18	16	21475.8562	21433.0563	21.3999	21475.8562	21433.0563	21.3999	{ 35}
22	4	8	11	10	13365.9352	13395.6602	29.7250	13365.9352	13395.6602	29.7250	{ 36}
22	2	10	14	11	15952.9171	15988.5449	11.8759	15952.9171	15988.5449	11.8759	{ 37}
26	8	9	15	14	20093.4903	20041.3105	52.1798	20093.4903	20041.3105	52.1798	{ 38}
26	2	5	13	12	14813.4230	14774.3246	39.0984	14813.4230	14774.3246	39.0984	{ 39}
27	5	8	12	11	14774.3246	14735.2262	39.0984	14774.3246	14735.2262	39.0984	{ 40}
27	8	11	11	10	14735.2262	14696.1278	39.0984	14735.2262	14696.1278	39.0984	{ 41}
29	0	4	11	10	12115.4383	12150.8502	35.4119	12115.4383	12150.8502	35.4119	{ 42}
30	1	3	18	17	20220.9368	20282.7530	61.8162	20220.9368	20282.7530	61.8162	{ 43}
31	2	11	9	7	10255.4467	10287.2895	15.9214	10255.4467	10287.2895	15.9214	{ 44}
31	2	11	18	14	20510.8934	20574.5789	15.9214	20510.8934	20574.5789	15.9214	{ 45}
31	4	9	13	11	15796.1053	15746.7440	24.6806	15796.1053	15746.7440	24.6806	{ 46}
32	1	7	13	11	14604.0099	14558.0305	22.9897	14604.0099	14558.0305	22.9897	{ 47}
32	4	10	12	10	14581.0202	14535.0408	22.9897	14581.0202	14535.0408	22.9897	{ 48}
32	4	10	18	15	21871.5304	21802.5612	22.9897	21871.5304	21802.5612	22.9897	{ 49}

Fig. 1. eCRAM processing of theoretical mass-to-charge ratios of the protein lysozyme. The raw input data (A) is processed and generates a list of candidate charge values (B) sorted on the basis of increasing error values, Δ_{ijkl} (i.e., "Diff" listed on Column 1). By searching the list in (B) with adjacency criteria, clusters of rows are found that form the "rowsets" for each charge carrying species illustrated in (C).

C: Calculated Row Sets by eCRAM

ROW SET AA [13,12]										
Diff	Peaks	Charges	RZa	RZb	delta_m	Ma	Mb	dM / dZ	{RowNum}	
1	0	3 13 12	14318.2452	14317.2374	1.0078	14305.1435	14305.1435	0.0000	{ 0}	
1	3	6 12 11	14317.2374	14316.2296	1.0078	14305.1435	14305.1435	0.0000	{ 1}	
1	6	9 11 10	14316.2296	14315.2218	1.0078	14305.1435	14305.1435	0.0000	{ 2}	
1	3	9 12 10	14317.2374	14315.2218	1.0078	14305.1435	14305.1435	0.0000	{ 4}	
1	0	6 13 11	14318.2452	14316.2296	1.0078	14305.1435	14305.1435	0.0000	{ 3}	
2	0	9 13 10	14318.2452	14315.2218	1.0078	14305.1435	14305.1435	0.0000	{ 7}	
ROW SET BA [13,12]										
Diff	Peaks	Charges	RZa	RZb	delta_m	Ma	Mb	dM / dZ	{RowNum}	
16	1	4 13 12	14604.0099	14581.0202	22.9897	14305.1429	14305.1430	0.0001	{ 25}	
16	4	7 12 11	14581.0202	14558.0305	22.9897	14305.1430	14305.1431	0.0001	{ 26}	
16	7	10 11 10	14558.0305	14535.0408	22.9897	14305.1431	14305.1431	0.0000	{ 27}	
32	4	10 12 10	14581.0202	14535.0408	22.9897	14305.1430	14305.1431	0.0000	{ 48}	
32	1	7 13 11	14604.0099	14558.0305	22.9897	14305.1429	14305.1431	0.0001	{ 47}	
47	1	10 13 10	14604.0099	14535.0408	22.9897	14305.1429	14305.1431	0.0001	{ 71}	
ROW SET CB [19,18]										
Diff	Peaks	Charges	RZa	RZb	delta_m	Ma	Mb	dM / dZ	{RowNum}	
14	5	6 19 18	23392.6807	23426.5576	33.8769	22649.8111	22722.7864	72.9753	{ 23}	
37	6	11 18 16	23426.5576	23513.8045	43.6235	22722.7864	22888.2301	82.7219	{ 59}	
52	5	11 19 16	23392.6807	23513.8045	40.3746	22649.8111	22888.2301	79.4730	{ 79}	
ROW SET CA [13,12]										
Diff	Peaks	Charges	RZa	RZb	delta_m	Ma	Mb	dM / dZ	{RowNum}	
26	2	5 13 12	14813.4230	14774.3246	39.0984	14305.1438	14305.1438	0.0000	{ 39}	
27	5	8 12 11	14774.3246	14735.2262	39.0984	14305.1438	14305.1438	0.0000	{ 40}	
27	8	11 11 10	14735.2262	14696.1278	39.0984	14305.1438	14305.1438	0.0000	{ 41}	
53	5	11 12 10	14774.3246	14696.1278	39.0984	14305.1438	14305.1438	0.0000	{ 81}	
53	2	8 13 11	14813.4230	14735.2262	39.0984	14305.1438	14305.1438	0.0000	{ 80}	
80	2	11 13 10	14813.4230	14696.1278	39.0984	14305.1438	14305.1438	0.0000	{ 129}	

Fig. 1. (Continued).

represented by the variable Δ_{ijkl} :

$$\Delta_{ijkl} = \left| \left(\left[\frac{(m/z)_i}{(m/z)_j} \right] \cdot \left[\frac{k}{l} \right] - 1 \right) \right| \quad (5)$$

For an ideal spectrum, the value for Δ_{ijkl} will be zero for the correct charge values k and l and their corresponding peaks i and j . However, this criterion cannot be applied alone. Uncertainties in the experimental measurements and/or spectral peak selection could result in physically realistic solutions having non-zero Δ_{ijkl} values. Conversely, non-physical solutions could have low values of Δ_{ijkl} . This is apparent when one recognizes that the ratios of integer charges are not all unique. For instance, the ratio of charges 18 and 15 is the same as the ratio of charges 12 and 10.

A valid combination (i, j, k, l) will always occur amongst a set of other related combinations (rows). In order to eliminate non-physical solutions, adjacency criteria are applied. The nature of this adjacency depends on the resolution of the original spectral data. In a low-resolution spectrum, the m/z peak positions occur in a set corresponding to a sequence of charge values for the same mass value. On the other hand, in a high-resolution spectrum, the peaks in a spectral cluster correspond to different isotopes of the same pair of charge values.

Consider two "rows", A and B, with their attributes denoted thus:

$$i_A, j_A, k_A, l_A, \quad [\Delta_{ijkl}]_A \\ i_B, j_B, k_B, l_B, \quad [\Delta_{ijkl}]_B$$

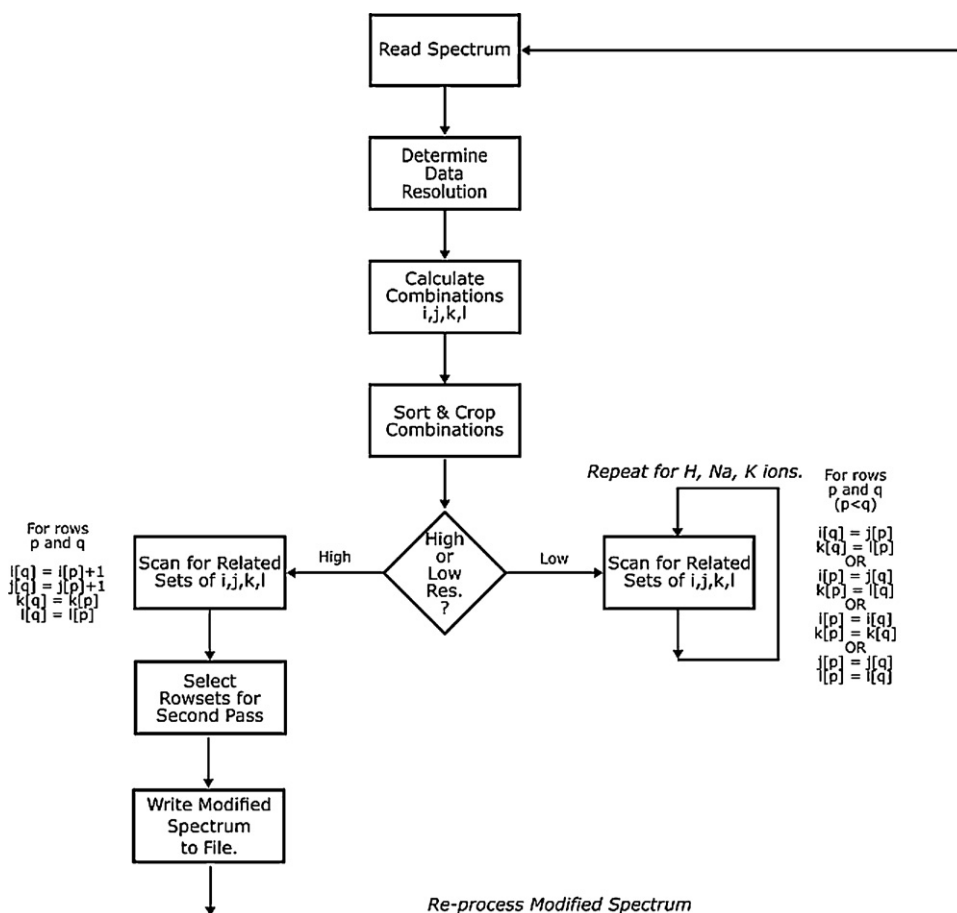
These two "rows" (that is, rows of the printout such as in Fig. 1B) are part of a "row set" if they satisfy certain adjacency criteria (see below). A "row set" is a coherent clump of peak identifiers and associated charge values. For low resolution, a row set will be comprised

of multiple charge combinations. For high-resolution data, a row set will be comprised of isotope peaks for the same charge combination. A valid row set represents a physically realistic solution that is internally consistent. In the analysis of the low-resolution spectra shown in Fig. 1C, the row set labelled AA is comprised of spectral peaks labelled 0 (1101.403480), 3 (1193.103120), 6 (1301.475420) and 9 (1431.522180) and their assigned charge values of 13, 12, 11 and 10. The consistency of the mass difference values (delta_m) and the Diff column adds further weight to the argument that this "row set" represents a physically realistic solution.

Scanning for combinations of peaks and charge values that form part of a "row set" involves scanning the combinations (i, j, k, l) that satisfy the adjacency conditions. In a low-resolution spectra, the adjacency condition requires one of the following to be true for the pair: ($i_B = j_A$ and $k_B = l_A$), ($i_A = j_B$ and $k_A = l_B$), ($i_A = i_B$ and $k_A = k_B$), or ($j_A = j_B$ and $l_A = l_B$). For the high-resolution mode, the adjacency condition requires $k_B = k_A$ and $l_B = l_A$.

There is no line drawn between "row set" boundaries. The rows cluster together into row sets based purely on their charge ratios, the adjacency criteria and their charge carrying species. The maximum difference values that are considered can be adjusted when the program is run. For both the low and high-resolution spectra, the best "row set" contains all the related charge values. Realistic solutions are observed to form into clusters corresponding to related combinations of charges. For example, a combination of charges 13 with 11 would be expected along with the combinations 13 with 12, and, 12 with 11. The row sets corresponding to unrealistic solutions may still be present in the final output and are discounted automatically or by the user input.

To analyze the isotopic effects in high-resolution spectra, the m/z values of the row sets of interest are corrected for the mass of the charge carrier and then the corrected spectrum is



Scheme 1. Flowchart for eCRAM processing of low and high-resolution mass spectral data.

completely re-processed. During the second pass, the program prompts the user to select a subset of the row sets to display as the final output. The molecular weights of the proteins (allowing for isotopic variations) are computed and printed out so as to highlight the correspondence of spectral peaks and associated protein mass.

The major processing steps performed by the eCRAM are shown in Scheme 1 and described below:

- *Step 1.* Load spectrum, normalize intensities, and determine data resolution (low versus high resolution).
- *Step 2.* Compute error values, Δ_{ijkl} , for all combinations of spectral peaks (i and j) and candidate charge values (k and l). These combinations of peaks and charges are referred to as rows (a user-defined lower and upper range can be used for the candidate charge values to minimize computing time). The error values, Δ_{ijkl} , are referred to as “Diff” for difference values in the program output and are scaled to a suitable range for display.
- *Step 3.* Sort the combinations (i, j, k, l) on increasing Δ_{ijkl} . A user-defined error range is typically selected (i.e., 0.03 for low resolution and 0.0001 for high-resolution data).
- *Step 4.* For each combination of i, j, k , and l , commencing at the smallest Δ_{ijkl} values, search for other combinations in the sorted list that form part of the same “row set”. The criteria for adjacency depend on the resolution of the spectral data.
- *Step 5.* For low-resolution spectra, display the best “row sets” for each candidate charge species. For high-resolution spectra, on the first pass, display the best “row sets” and allow for input of the row sets for subsequent re-processing for isotopic effects (Eq. (4)).

For high-resolution spectra, on the second pass, display the best “row sets” and allow for input of the row sets for final display.

- *Step 6.* For high-resolution spectra, on the second pass, calculate and display the peak registrations and corresponding protein mass values.

3.3. eCRAM processing of low-resolution electrospray mass spectral data

The example provided in Fig. 1 illustrates the application of eCRAM utilizing theoretical mass-to-charge ratios of the protein lysozyme [11]. This data set is for low-resolution ESI mass spectra and contains protonated as well as adducts of sodium and potassium ions. The list shown in Fig. 1A has 12 m/z values in ascending order with their corresponding row numbers. For example, row zero has a corresponding m/z value of 1101.4034. The intensity of all ions in this example is set to an arbitrary value of 100 to represent the theoretically calculated m/z values. The program recognizes that the set is for low-resolution data with one bunch of spectral peaks (i.e., no isotopic peaks), and computes m/z ratio values along with candidate charge ratio values. In this example, the lower and upper values of the matrix were 7 and 19 (i.e., a user-defined range minimizes the computing time).

The output shown in Fig. 1B shows the best 50 rows sorted on the basis of increasing error values, Δ_{ijkl} (i.e., “Diff” listed on Column 1). The mass of charge carrying species listed as “delta_m” on Column 8 of Fig. 1B is calculated from Eq. (3). The correction factor (Eq. (4)) for isotopic correlation of high-resolution data is represented as $|dM|/|dZ|$ under Column 11, which is the same value

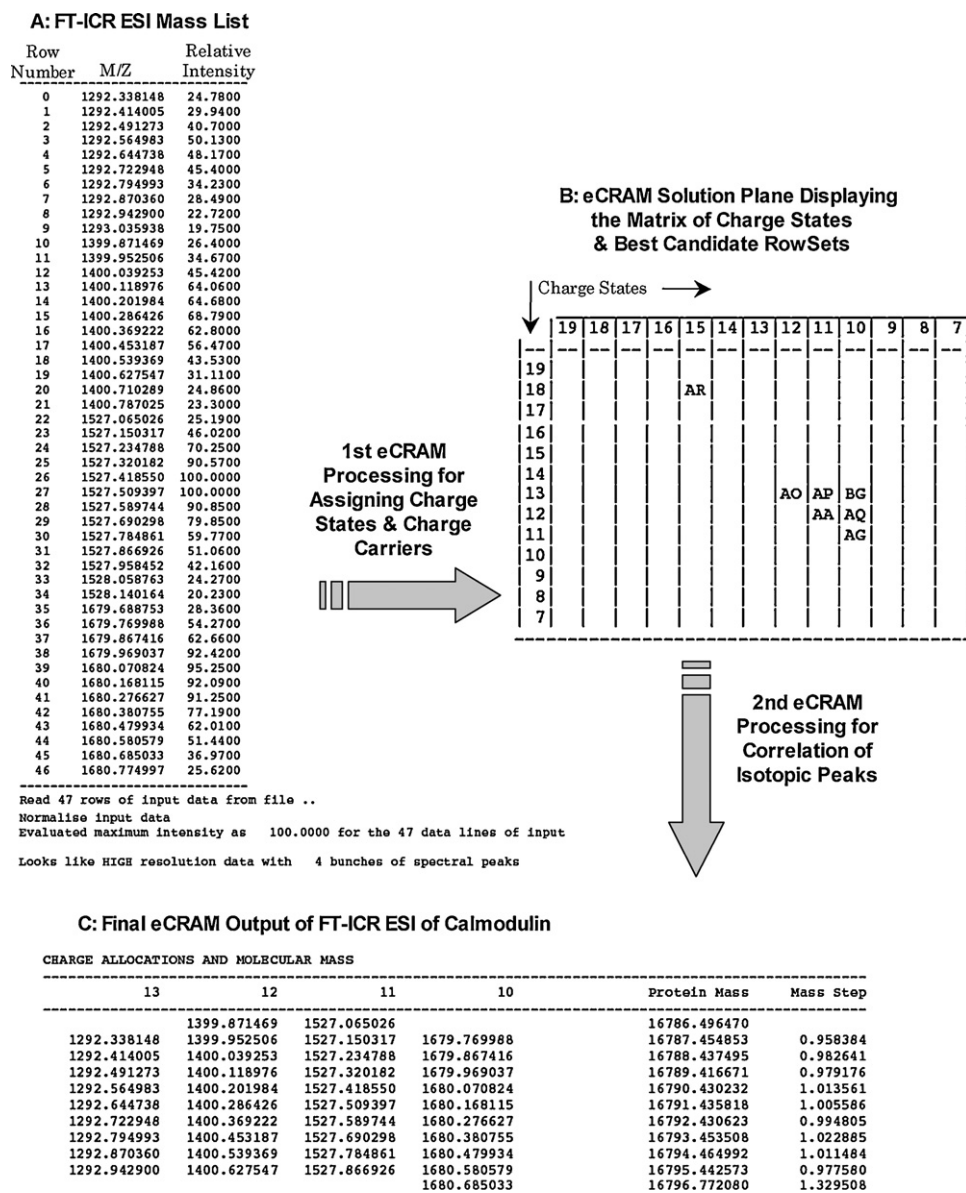


Fig. 2. eCRAM processing of high-resolution FT-ICR electrospray mass spectrum of calmodulin. The rowsets within the triangular region bounded by (AO, BG and AG) are selected for second stage processing.

as “delta_m” in this example of low-resolution data processing (Fig. 1B).

The program assembles the row sets in the following way. By starting with the best row (smallest Diff value), the eCRAM algorithm searches down the list for other rows that satisfy the adjacency criteria required to join the current “row set”. The program then returns to the top of the list and selects the next best, non-assigned row and repeats the process. A number of filtering and merging steps are performed to obtain the rowset depicted in Fig. 1C. The assembly of the row set AA depicted in Fig. 1C, can be explained by reading down the “RowNum” in Fig. 1B (last column):

- RowNum 0 – the row (0, 3, 13, 12) is the nucleation point for this “row set” as it occurs first in the row list (best difference value).
- RowNum 1 – the row (3, 6, 12, 11) satisfies the adjacency criteria with RowNum 0 (i.e., 3 = 3 and 12 = 12), and also RowNum 2 and 3.
- RowNum 2 – the row (6, 9, 11, 10) satisfies the adjacency criteria with RowNum 1 (i.e., 6 = 6 and 11 = 11), and also RowNum 3 and 4.
- RowNum 3 – the row (0, 6, 13, 11) satisfies the adjacency criteria with RowNum 2 (i.e., 6 = 6 and 11 = 11), and also RowNum 1 and 7.

- RowNum 4 – the row (3, 9, 12, 10) satisfies the adjacency criteria with RowNum 0 (i.e., 3 = 3 and 12 = 12), and also RowNum 2 and 7.
- RowNum 5 – the row (3, 9, 18, 15) is not adjacent to any row set members and appears to be an integer multiple of the charge ratio 12:10.
- RowNum 6 – the row (0, 5, 19, 17) is not adjacent to any row set members and the delta_m is not an expected value.
- RowNum 7 – the row (0, 9, 13, 10) satisfies the adjacency criteria with RowNum 3 (i.e., 0 = 0 and 13 = 13) and RowNum 4 (i.e., 9 = 9 and 10 = 10).

eCRAM has matched all protonated species as part of the row set AA. When looking for sodium species (22.9897), the nucleation point is the row (1, 4, 13, 12), and the adjacency criteria result in the rows forming the row set BA (Fig. 1C) from peaks 1, 4, 7, 10 with charges of 13, 12, 11 and 10. Similarly for potassium (39.0984), the row set CA (Fig. 1C) is comprised of peaks 2, 5, 8 and 11 with corresponding charges of 13, 12, 11 and 10. Note that although the row set CB satisfies the adjacency criteria, inspection of the delta_m values discounts it as a physically unrealistic solution.

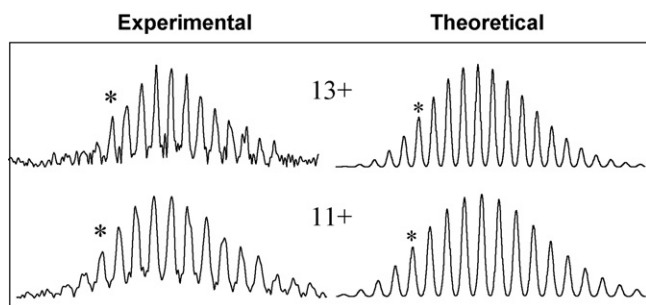


Fig. 3. Correlation of the experimental and theoretical isotopic peaks for +13 and +11 ions of calmodulin by the eCRAM.

Note that the values of charge carrying species (i.e., proton, sodium and potassium) are consistent throughout various row sets. For instance, the charge carrying species calculated for all six sets of charges of row set AA is 1.0078 (proton). Similarly, for the row set BA, the charge carrying species is 22.9897 (sodium) and for the row set CA, a value of 39.0984 (potassium) is calculated. This process of identifying the charge carrying species, purely from the ratios of

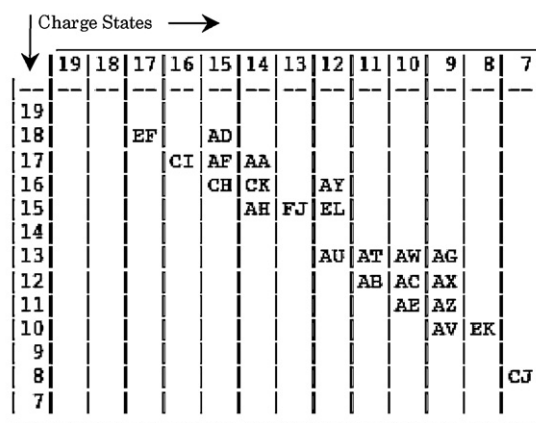
charge states, is a unique feature of the CRAM. Values for Δ_{ijkl} (Eq. (5)) listed under “Diff” are larger for row sets BA and CA when compared to the row set AA. This is expected and is due to the increase in the mass of charge carrying species and the assumption of Eq. (3) ($M > z_i m_A$ or $z_j m_A$).

As described above in the description of the eCRAM algorithm, non-realistic solutions could also be found (for example, the ratio of charges 18 and 15 is the same as the ratio of charges 12 and 10). For the lysozyme data, Fig. 1C shows the row set CB with acceptable error values. This set matches data for charge states 19 and 18, 18 and 16, and 19 and 16, while missing other key charge states of 18 and 17, or 19 and 17. Additionally, the value of charge carrying species is not consistent for each set of charges. These two features identify that the row set CB is not a valid realistic solution and could be removed.

3.4. eCRAM processing of high-resolution electrospray mass spectral data

Processing of high-resolution mass spectral data occurs in two steps [12]. In the first instance, the charge state values of ions

A: eCRAM Solution Plane Displaying the Matrix of Charge States & Best Candidate RowSets



B: 2nd eCRAM Processing for Correlation of Isotopic Peaks

CHARGE ALLOCATIONS AND MOLECULAR MASS							
	13	12	11	10	9	Protein Mass	Mass Step
		714.394302	779.246726	857.072612		8560.637835	
659.594461	714.477019	779.337797	857.171580	952.302124		8561.634506	0.996671
659.672138	714.559723	779.428320	857.270557	952.412209		8562.630213	0.995707
659.748504	714.642665	779.518943	857.370202	952.522463		8563.624943	0.994730
659.825456	714.725783	779.609544	857.470326	952.633203		8564.623404	0.998461
659.902350	714.808797	779.700349	857.570216	952.744634		8565.622689	0.999285
659.979543	714.892528	779.791643	857.670461	952.857078		8566.628081	1.005392
660.056646	714.976234	779.883396	857.771175	952.968321		8567.632965	1.004884
660.133886	715.060261	779.974690	857.871427	953.079256		8568.636488	1.003523
660.211243	715.144068	780.066235	857.972169	953.192745		8569.645916	1.009428
660.289509				953.305739		8570.671559	1.025643

CHARGE ALLOCATIONS AND MOLECULAR MASS						
	17	16	15	14	Protein Mass	Mass Step
727.754594	773.175045	824.652201	883.486267		12354.683605	
727.809125	773.237850	824.719175	883.559322		12355.673427	0.989822
727.868258	773.298050	824.784275	883.629219		12356.654307	0.980880
727.926523	773.360287	824.849754	883.699709		12357.643142	0.988835
727.983267	773.421284	824.916583	883.771239		12358.629256	0.986114
728.043451	773.483206	824.982471	883.841123		12359.624400	0.995144
728.103003	773.545256	825.048383	883.914328		12360.629083	1.004683
728.161258	773.605879	825.115062	883.985178		12361.617180	0.988097
728.219739	773.670374	825.183746	884.057970		12362.636042	1.018861
728.280047		825.252235	884.126985		12363.654055	1.018013

Fig. 4. eCRAM processing of the combined high-resolution FT-ICR electrospray mass spectra of ubiquitin and cytochrome *c* as an example of mixtures. Two distinct rowsets (i.e., mixture of two proteins) within the triangular regions bounded by (AU, AG and AV) and (CI, AA, AH) are selected for second stage processing.

are calculated along with the mass (m_A) and subsequent identification of the charge carrying species (Eq. (3)). In the following steps, the isotopic peaks of the multiply charged ions that share a common isotopic composition are correlated (Eq. (4)). These are illustrated with high-resolution mass spectra of the protein calmodulin (Figs. 2 and 3).

The list in Fig. 2A contains 47 rows of m/z and intensity values. The program first recognizes that the list represents a set of high-resolution data with four bunches of spectral lines. The row sets are then identified from the ratios of m/z values and by comparing those to ratios of integers. The matrix in Fig. 2B shows seven possible row sets, all within an acceptable error value, $\Delta_{ijkl} = 0.0005$ (Eq. (5)). The row set AR (corresponding to charge state 18, 15) does not correlate with other Row Sets, and is eliminated at this stage. The other six row sets (i.e., AO, AP, BG, AA, AQ, AG) represent correlation of ion charges 13, 12, 11 and 10. The charge carrying species is identified as a "proton" (as described above for low-resolution data).

For the correlation of the isotopic peaks, m/z values are adjusted and eCRAM generates a new list of m/z values based on the correction applied from Eq. (4) (i.e., the mass and the numbers of charge carrying species associated with each ion). eCRAM applies the algorithm to the new m/z list and generates the final output (Fig. 2C) that correlates isotopic peaks for all ions. Note in this example, the first isotope peak of charge 13 (m/z 1292.338148), the second isotope peaks of charge 12 and 11 (i.e., m/z 1399.952506 and 1527.150317) and the first isotope peak of charge 10 (m/z 1679.769988) are correlated. The protein mass (i.e., neutral mass) is calculated from the average of m/z values multiplied by their corresponding charges and subtracting the mass of charge carrying species (i.e., proton in this example). The protein mass based on the lightest isotopic peak is 16786.496470 as listed on the first row of Column 5 (Fig. 2C).

The eCRAM correlation of isotopic peaks for high-resolution ESI data is also helpful for matching experimental and theoretical isotopic patterns [14]. In the case of compounds where the elemental composition is not known, an elemental composition is assumed ($C_aH_bO_cN_dS_e$). Correlation of the experimental and theoretical isotopic peaks for +13 and +11 ions of calmodulin by the eCRAM is shown in Fig. 3. Similar results for processing of both low and high-resolution ESI data of five proteins, ubiquitin, lysozyme, cytochrome *c*, transthyretin (TTR) and calmodulin were obtained by the eCRAM. In all these cases, the eCRAM algorithm correctly identified the ion charge and correlated isotopic peaks.

3.5. eCRAM processing of high-resolution electrospray mass spectral data for mixtures

To evaluate the eCRAM algorithm for processing of complex sample mixtures, mass spectral data for two proteins, ubiquitin and cytochrome *c* were combined. The simulated data for mixtures contained high-resolution m/z values for charge states 9 to 13 of ubiquitin, and charge states 14 to 17 for cytochrome *c*. The eCRAM algorithm successfully processed the mixture data by correctly calculating the charge states of the two proteins, and subsequently correlated their charge states with m/z values of their isotopic peaks. The eCRAM output with charge allocations and molecular mass of ubiquitin and cytochrome *c*, from processing the proteins as a mixture, are presented in Fig. 4. Note in this case, the matrix (Fig. 4A) displays two distinct triangular regions bounded by (AU, AG and AV) and (CI, AA, AH) that represent the charge states and their associated row sets for the two proteins.

The eCRAM algorithm was evaluated further by including an impurity in the form of an overlapping peak (data not shown). Mass

spectral data for +18 charge state of transthyretin (TTR) were added to the mass list of ubiquitin and cytochrome *c* mixture (described above). In this instance, 10 extra peaks associated with TTR ranging from m/z 772.599554 to 773.099633, partially overlapped with +16 peaks of cytochrome *c* ranging from m/z 773.114083 to 773.795056. The eCRAM algorithm once again identified and correlated correct charge states with m/z values associated with ubiquitin and cytochrome *c*, and the presence of the impurity did not disturb the processing of the data.

4. Conclusions

The eCRAM computer program and the processing of both low and high-resolution data were described. In addition, methodologies for the selection of real solutions and elimination of non-realistic solutions were provided. Mass spectral data for a range of proteins including mixtures, were successfully processed by the eCRAM algorithm, and examples were provided for lysozyme, calmodulin, and a mixture containing ubiquitin and cytochrome *c*. The eCRAM demonstrated that both the ion charge and the mass of charge carrying species are calculated without *a priori* knowledge and assumption of the nature of charge carrying species, and more importantly, the charge carrying species is not assumed to be *uniform* across an electrospray mass spectrum. The isotope correlation feature of the eCRAM is especially useful for quantitative applications where cross-correlation of mass spectral data with theoretical isotopic distribution is required, particularly for low signal-to-noise data and where elemental compositions are not known. For accurate processing of the high-resolution mass spectral data by eCRAM some user-defined input for selection of a set of solutions is recommended. The eCRAM algorithm is currently being optimized for availability on the World Wide Web. Options for a standalone version are also being explored with provisions to provide the program code to non-commercial users.

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References

- [1] J.B. Fenn, M. Mann, C.K. Meng, S.F. Wong, G.M. Whitehouse, *Science* 246 (1989) 64.
- [2] F.W. McLafferty, *Acc. Chem. Res.* 27 (1994) 379.
- [3] T.R. Covey, R.F. Bonner, B.I. Shushan, J. Henion, R.K. Boyd, *Rapid Commun. Mass Spectrom.* 11 (2) (1988) 249.
- [4] M. Mann, C.K. Meng, J.B. Fenn, *Anal. Chem.* 61 (1989) 1702.
- [5] B.B. Reinhold, V.N. Reinhold, *J. Am. Soc. Mass Spectrom.* 3 (1992) 207.
- [6] J.J. Hagen, C.A. Monnig, *Anal. Chem.* 66 (1994) 1877.
- [7] M.W. Senko, S.C. Beu, F.W. McLafferty, *J. Am. Soc. Mass Spectrom.* 6 (1995) 52.
- [8] M.W. Senko, S.C. Beu, F.W. McLafferty, *J. Am. Soc. Mass Spectrom.* 6 (1995) 229.
- [9] Z. Zhang, A.G. Marshall, *J. Am. Soc. Mass Spectrom.* 9 (1998) 225.
- [10] D.M. Horn, R.A. Zubarev, F.W. McLafferty, *J. Am. Soc. Mass Spectrom.* 11 (4) (2000) 320.
- [11] L. Chen, S.K. Sze, H. Yang, *Anal. Chem.* 78 (14) (2006) 5006.
- [12] L. Chen, H. Yang, *J. Am. Soc. Mass Spectrom.* 19 (2008) 46.
- [13] S.D. Maleknia, K.M. Downard, *Anal. Chem.* 77 (2005) 111.
- [14] S.D. Maleknia, K.M. Downard, *Int. J. Mass Spectrom.* 246 (2005) 1.
- [15] M.T. Olson, A.L. Yergey, *J. Am. Soc. Mass Spectrom.* 20 (2009) 295.
- [16] J. Meija, *Anal. Bioanal. Chem.* 385 (2006) 486.
- [17] F. Former, L.J. Foster, S. Toppo, *Curr. Bioinform.* 2 (2007) 63.
- [18] M. Wehofsky, R. Hoffmann, *J. Mass Spectrom.* 37 (2002) 223.
- [19] J. Fernandez-de-Cossio, L.J. Gonzales, Y. Satomi, L. Betancourt, Y. Ramos, V. Huerta, V. Besada, G. Padron, N. Minamino, T. Takao, *Rapid Commun. Mass Spectrom.* 18 (2004) 2465.
- [20] S.D. Maleknia, N. Reixach, J.N. Buxbaum, *FEBS J.* 273 (2006) 5400.